## PHARM. D DEGREE EXAMS THIRD YEAR PAPER II – PHARMACEUTICAL ANALYSIS

Q.P. Code: 383814

Time: 3 hours Maximum: 70 marks

I. Elaborate on : (2x20=40)

1. a) With a neat labeled diagram, explain the principle and instrumentation of HPLC.

- b) Write principle, stationary phase, mobile phase, development techniques and applications of paper chromatography.
- 2. a) Describe the theory of IR absorption. With a neat labeled diagram, explain the instrumentation of IR spectroscopy. Explain its applications in qualitative analysis of pharmaceuticals.
  - b) Write the principle involved in NMR and Mass spectroscopy.

II. Write notes on: (10x3=30)

- 1. Define chromophore, auxochrome and hypsochromic shift.
- 2. Write the applications of DSC and DTA in Pharmaceutical Analysis.
- 3. Define GLP, ICH and Validation.
- 4. What is base peak? List the different types ions produced in mass spectra.
- 5. List at least three differences between TLC and HPTLC.
- 6. Discuss briefly about ESR and its application.
- 7. Write the differences between flame emission and atomic absorption spectroscopy.
- 8. Define electrode potential. Give example of reference and indicator electrode used in potentiometry.
- 9. What do you know about dropping mercury electrode? Give any two advantages and disadvantages.
- 10. Write a brief note on quenching.

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