B.PHARM. EXAMINATION SECOND YEAR

PAPER II – PHARMACEUTICAL ANALYSIS & PHYSICAL CHEMISTRY

Q.P. Code: 564257

Time: Three hours Maximum: 100 marks

Answer All Questions

Answer Section-A and B in separate Answer Book SECTION-A

(Pharmaceutical Analysis)

I. Essay: $(2 \times 10 = 20)$

- 1. a) Give an account on theory of non-aqueous titration and also mention about its application in pharmacy.
 - b) Write note on Iodimetry and Iodometry titration.

II. Short notes: $(4 \times 5 = 20)$

- 1. Write notes on error
- 2. Law of mass action
- 3. Assay of calcium gluconate
- 4. Volhard's methods

III. Short answers: $(5 \times 2 = 10)$

- 1. Precision
- 2. Indicators
- 3. Co-precipitation
- 4. Iodine value
- 5. Ligands

SECTION-B

(Physical Chemistry)

I. Essay: $(2 \times 10 = 20)$

1. a) Give an account on determination methods of osmotic pressure.

b) Factors affecting adsorption and application of adsorption in pharmacy.

II. Short notes: $(4 \times 5 = 20)$

- 1. Phase rule and its application
- 2. Hess's law of constant heat of summation
- 3. Entropy and its significance
- 4. Explain the principle involved in polarimetry

III. Short answers: $(5 \times 2 = 10)$

- 1. Third law of thermodynamics
- 2. Adsorption isotherm
- 3. Rate of reaction
- 4. Auto catalyst
- 5. Molecularity of reaction
