

[LB 4258]

AUGUST 2012

Sub. Code: 4258

SECOND YEAR B.PHARM. EXAM

Paper III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code : 564258

Time: Three hours

Maximum: 100 Marks

(180 Min) Answer ALL questions in the same order.

I. Elaborate on:

Pages Time Marks
(Max.)(Max.)(Max.)

- | | | | |
|---|----|----|----|
| 1. Define and classify alkaloids with examples. Discuss the chemistry and pharmacological activity of ephedrine. | 19 | 33 | 20 |
| 2. Write the definition, reaction, mechanism and applications of (i) Beckmann rearrangement and (ii) Birch reduction. | 19 | 33 | 20 |

II. Write notes on:

- | | | | |
|--|---|---|---|
| 1. Write a brief note on Absolute configuration. | 3 | 8 | 5 |
| 2. Explain Clemmenson's reduction in detail. | 3 | 8 | 5 |
| 3. (i) Fischer indole synthesis (ii) Skraup synthesis. | 3 | 8 | 5 |
| 4. Explain the oxidative uses of lead tetraacetate. | 3 | 8 | 5 |
| 5. Give any two methods of preparation of Oxazole and Thiophene. | 3 | 8 | 5 |
| 6. Explain conformational analysis of ethane. | 3 | 8 | 5 |
| 7. Write short notes on Atropine and related alkaloids. | 3 | 8 | 5 |
| 8. Write the electrophilic substitution reactions of Pyrrole. | 3 | 8 | 5 |

III. Short Answers:

- | | | | |
|---|---|---|---|
| 1. Define racemic modification. | 1 | 5 | 2 |
| 2. What are enantiomers. | 1 | 5 | 2 |
| 3. Write the definition and reaction for Meerwein Pondoiff reduction. | 1 | 5 | 2 |
| 4. Give any two methods of preparation of pyrimidine. | 1 | 5 | 2 |
| 5. What is dehydrogenation. | 1 | 5 | 2 |
| 6. What are flavonoids. | 1 | 5 | 2 |
| 7. Write any two uses of Selenium oxide. | 1 | 5 | 2 |
| 8. Define alkaloids and give examples. | 1 | 5 | 2 |
| 9. Write the structure of i) Vitamin D ii) Ephedrine | 1 | 5 | 2 |
| 10. Give any two reactions of thiophene. | 1 | 5 | 2 |

[LC 4258]

FEBRUARY 2013

Sub. Code: 4258

SECOND YEAR B.PHARM. EXAM

Paper III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code : 564258

Time : Three hours

Maximum: 100 Marks

(180 Min)

I. ELABORATE ON:

(2x20=40)

1. a) Write in detail about asymmetric synthesis(10)
b) Describe the stereochemistry of cyclic compounds(10)
2. a) Describe the chemistry and constitutions of ephedrine(15)
b) Give the structure and five reactions of atropine(5)

II. SHORT NOTES

(8x5=40)

1. Write the synthesis of furan
2. Describe the Clemmenson's reduction with suitable reaction
3. Classify heterocyclic compounds with example
4. Explain the Beckmann rearrangement with mechanism
5. Classify flavonoids with example
6. Describe the chemistry of digitoxin
7. Give the chemical reactions of papaverine
8. Describe the chemistry and uses of vitamin-E

III. SHORT ANSWERS

(10x2=20)

1. Add a note on sequence rule
2. Write any two reactions given by pyridine
3. Give two examples for the oxidative reactions of periodic acid
4. Describe the reactions of thiazole
5. Give the uses of metal hydride
6. Write the preparation of citral
7. Discuss the chemistry of uric acid
8. Write the chemistry of folic acid
9. Sketch the structures of caffeine and theobromine
10. Draw the structure and chemical name of thiamine

(LD 4258)

AUGUST 2013

Sub. Code: 4258

SECOND YEAR B.PHARM. EXAM

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three Hours

Maximum: 100 marks

I. Elaborate on:

(2X20=40)

1. a) What are conformational isomers and write in detail about conformational analysis.
b) Describe the modern theories of double bond with suitable examples.
2. Elucidate the structure of uric acid. Discuss the inter relationship between caffeine, Theophylline and theobromine.

II. Write notes on:

(8X5=40)

1. Explain the Schmidt rearrangement with mechanism?
2. Give the chemical reactions of atropine.
3. Describe the chemistry and uses of Vitamin K.
4. Write a brief note on Asymmetric synthesis.
5. Give any two methods of preparation and any three reactions of Furan.
6. Write a brief note on catalytic hydrogenation.
7. What is chirality? Write their significance.
8. Write a note on Walden Inversion.

III. Short Answers on:

(10X2=20)

1. Write any two reactions given by pyrimidine.
2. What is Darzens reactions?
3. Give any two reactions of ephedrine.
4. Give any two reaction of indole.
5. Write any two uses of Lithium aluminium hydride.
6. Define alkaloids and give examples.
7. Write the structure and numbering of heterocyclic compounds pyrimidine and pyrrole.
8. Medicinal uses of Flavanoids.
9. Define racemic modifications.
10. Write a note on plane of symmetric.

(LE 4258)

FEBRUARY 2014

Sub. Code: 4258

SECOND YEAR B.PHARM. EXAM
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY
Q.P. Code: 564258

Time: Three Hours

Maximum: 100 marks

I. Elaborate on:

(2X20=40)

1. a) Describe the absolute and relative configuration.
b) Stereospecific and stereoselective synthesis.
2. Explain the chemistry and constitution of citral.

II. Write notes on:

(8X5=40)

1. Clemmenson's reduction.
2. Reaction of furan.
3. Stereochemistry of cyclic compounds.
4. Chemistry of papaverine.
5. Relationship between menthol and thymol.
6. Schmidt rearrangement.
7. Synthesis of thiophene.
8. Structure and uses of Vitamin B₁ and E.

III. Short Answers on:

(10X2=20)

1. Define glycoside.
2. Classify terpenoids.
3. Structure and uses of digoxin.
4. Define plane of symmetry.
5. Fischer projection.
6. Structure of hespiridine.
7. Two reaction of pyridine.
8. Structure of atropine and ephedrine.
9. Darzen's reaction.
10. Two synthesis of pyrimidine.

SECOND YEAR B.PHARM. DEGREE EXAMINATION**Paper III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY****Q. P. Code: 564258****Time: Three Hours****Maximum: 100 Marks****Answer All Questions****I. Essay Questions: (2 x 20 = 40)**

1. a) What is Walden Inversion explain in detail the mechanism of reaction and factors affecting Walden inversion.
b) Give important methods of preparation and chemical reactions of Pyridine.
2. a) Give the structural elucidation of Atropine.
b) Give the definition, mechanism of reaction and applications of Meerwein-Ponndorf reaction and Schmidt reaction

II. Short Notes: (8 x 5 = 40)

1. Explain Catalytic hydrogenation.
2. Explain the various methods of nomenclature of Geometrical isomers.
3. Write the chemistry of vitamin A.
4. Elucidate the structure of Caffeine.
5. Explain the chemistry of Lanatosides.
6. Give the Halogenation, Coupling reaction and Reduction reaction of Furan.
7. Write the synthesis of Pyrimidine.
8. Explain any five methods of resolution of racemic mixture.

III. Short Answers: (10 x 2 = 20)

1. Give the application of mercuric acetate.
2. Write the structure and medicinal uses of Digoxin and Theophylline.
3. Write the Friedel-Crafts reaction and Coupling reaction of Pyrrole.
4. Write any two reactions of Acridine.
5. Give the structure and medicinal uses of Thymol and Hesperidin.
6. Define Enantiomer and Diastereoisomer.
7. Define Clemmensen reaction with example.
8. Write any method of synthesis of phenothiazine.
9. List out the Alkaloidal reagents used for identification.
10. Classify glycosides based on the Aglycones present in it.

(LG 4258)

FEBRUARY 2015

Sub. Code: 4258

**SECOND YEAR B.PHARM. EXAMINATION
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY**

Q.P. Code: 564258

Time: Three hours

Maximum: 100 marks

I. Essay: (2 x 20 = 40)

1. a) Write in detail about determination of configuration of geometrical isomerism.
b) Write notes on Beckmann rearrangement.
2. Explain the chemistry and constitution of Menthol and Vitamin A.

II. Short notes: (8 x 5 = 40)

1. Metal hydride reduction
2. Reactions of indole
3. Sequence rule for assigning configuration of optical isomers
4. Give the important electrophilic substitution reactions of pyrrole
5. Chemistry of caffeine
6. Structure and uses of citral and thymol
7. Synthesis of isoxazole
8. Chemistry of sennosides

III. Short answers: (10 x 2 = 20)

1. What is chirality?
2. What is Stereomutation?
3. Give the Structure and uses of folic acid
4. Define diastereomers
5. Explain the D and L system in optical isomerism
6. What is Stereoselective synthesis?
7. What is Catalytic dehydrogenation?
8. Give two methods of synthesis of imidazole
9. Classify flavonoids
10. Structure and uses of azepines

B.PHARM. DEGREE EXAMINATION**SECOND YEAR****PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY****Q.P. Code: 564258****Time : Three Hours****Maximum : 100 marks****Answer All Questions****I. Essay:****(2 x 20 = 40)**

1. a) What is cis-trans isomerism? Explain the E-Z system of nomenclature.
b) Explain the various methods of determination of configuration of geometric isomers.
2. a) Define and classify alkaloids. Explain the chemistry of Papaverine.
b) What are cardiac glycosides? Explain the stereochemistry of cardiac glycosides.

II. Short notes :**(8 x 5 = 40)**

1. Define resolution of racemic modification. Explain the method of resolution by formation of diastereomers.
2. Explain conformational analysis of monosubstituted cyclohexane.
3. Discuss the interrelation between caffeine, theophylline and theobromine.
4. What happens when
 - a) Pyridine is oxidized with perbenzoic acid.
 - b) Quinoline is treated with sulphuric acid.
 - c) Pyrrole is reduced with nickel.
 - d) Imidazole is treated with benzoyl chloride.
 - e) Furan is treated with maleic anhydride.
5. Explain optical isomerism in meso-tartaric acid.
6. Explain the chemistry of citral.
7. Classify flavonoids and give examples. Write a brief note on hesperidin.
8. Give reasons for the following.
 - a) Pyrrole undergoes electrophilic substitution at 2-position.
 - b) Pyridine undergoes electrophilic substitution at 3-position and nucleophilic substitution at 2- and 4-positions.

III. Short answers:**(10 x 2 = 20)**

1. What is Walden inversion?
2. What is 1,3-diaxial interaction?
3. What is Birch reduction?
4. Give the structure and use of (a) Thymol (b) Camphor.
5. Define plane of symmetry
6. Give one test for the identification of Atropine.
7. Write any two reactions of isoquinoline
8. What are configurational and conformational isomers?
9. Write any two uses of sodium borohydride
10. What is Schmidt rearrangement?

(LI 4258)

FEBRUARY 2016

Sub. Code: 4258

SECOND YEAR B.PHARM. EXAMINATION
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Essay: (2 x 20 = 40)

1. Define racemic modification. Explain the various methods of resolution of racemic modification.
2. Define and classify heterocyclic compounds with examples. Explain the nomenclature of heterocyclic compounds. Write the methods of preparation and reactions of quinoline.

II. Short notes: (8 x 5 = 40)

1. Explain the stability of the possible conformations of disubstituted cyclohexane.
2. Explain sequence rules giving examples.
3. Compare and contrast Clemmensen reduction and Wolf Kishner reduction.
4. a) Explain the basicity of Pyrrole and Pyridine.
b) Give any two electrophilic substitution reactions of furan.
5. Explain DL system of nomenclature. What are its disadvantages?
6. Discuss the chemistry and pharmacological activity of Ephedrine.
7. Define and classify Terpenoids. Add a note on the chemistry and uses of Thymol.
8. What are the applications of (a) Selenium oxide (b) Lithium Aluminium Hydride?

III. Short answers: (10 x 2 = 20)

1. Why Nitrobenzene is mostly used as oxidizing agent in Skraup synthesis?
2. What is 1, 3-diaxial interaction?
3. Write the structure and numbering of any two 5-membered heterocyclic compounds containing two similar hetero atoms.
4. Give the structure and numbering of (a) Isoquinoline (b) Phenothiazine.
5. Give one test for identification of the Steroidal Nucleus.
6. What are Sennosides?
7. Write any two reactions of Imidazole.
8. Define Enantiomers and Diastereomers.
9. Write the structure and uses of Vitamin B₁.
10. What do you mean by centre of symmetry? Explain with an example.

(LJ 4258)

AUGUST 2016

Sub. Code: 4258

**SECOND YEAR B.PHARM. EXAMINATION
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY**

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Essay: (2 x 20 = 40)

1. What is Conformational Analysis? With the help of a potential energy diagram explain the stability of the various possible conformations for ethane and 1, 2-dichloroethane.
2. a) What are Purines? Elucidate the structure of uric acid.
b) Explain the interrelation between the Xanthine alkaloids.

II. Short notes: (8 x 5 = 40)

1. Define Optical activity? What are the various optically active and inactive forms possible in compounds containing two different chiral carbons?
2. a) Explain the modern theory of double bonds.
b) E-Z system of nomenclature.
3. List out the methods used to reduce carbonyl compounds to hydrocarbons. Write the reaction, mechanism and applications of any one method.
4. What happens when:
 - a) Pyrrole is treated with Maleic anhydride.
 - b) Thiophene is treated with Raney Nickel.
 - c) Indole is treated with chloroform and potassium hydroxide.
 - d) Quinoline is oxidised with potassium permanganate.
 - e) Pyrazole is treated with acetic anhydride.
5. Discuss the stereochemistry of cardiac glycosides.
6. Explain the chemistry of Vitamin B₆.
7. Write any two methods of preparation and any three reactions of pyrimidine.
8. Give the reaction, mechanism and salient features of Beckmann rearrangement.

III. Short answers: (10 x 2 = 20)

1. What are the elements of symmetry?
2. Write any two uses of lead tetraacetate.
3. What is Darzen's reaction?
4. Give the structure of (a) Atropine (b) Digoxin
5. Define stereo-selective synthesis.
6. What is Chichibabin reaction?
7. Define flavonoids and give examples.
8. Write the halogenation reaction of quinoline.
9. Define asymmetric synthesis.
10. What is Meerwin-Pondorf Verley reduction?

(LK 4258)

FEBRUARY 2017

Sub. Code: 4258

**B.PHARM. EXAMINATION
SECOND YEAR
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY**

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on: **(2 x 20 = 40)**

1. What are Alkaloids? Classify with examples and add the chemistry of Atropine.
2. Briefly explain on the following reactions.
 - a) Clemmenson's reduction
 - b) Beckmann rearrangement
 - c) Meerwin – Pondoiff reduction
 - d) Birch reduction.

II. Write notes on: **(8 x 5 = 40)**

1. Write the applications of periodic acid and mercuric acetate.
2. Explain the chemistry of Vitamin D.
3. Discuss the stereochemistry of cyclic compounds.
4. Briefly explain on the conventions used in stereochemistry.
5. Give an account on synthetic method of preparation and reactions of
 - a) Acridine
 - b) Phenothiazine
6. Explain the chemistry and uses of camphor.
7. Discuss the chemistry of Folic acid.
8. Explain the following reactions:
 - a) Friedel-craft acylation of thiophene
 - b) Reimer-Tiemann Formylation of indole.

III. Short answers on: **(10 x 2 = 20)**

1. Write the structures of Xanthine bases.
2. What is meant by chirality? Give an example.
3. What happens when methyl substituted pyrazole is oxidized by potassium permanganate?
4. Give the classification of flavonoids.
5. What is alternating axis of symmetry?
6. Give the structure and numbering of digoxin.
7. What is meant by catalytic hydrogenation?
8. Define isomers and isomerism.
9. Write the pharmacological activity of ephedrine.
10. Write the molecular formula for monoterpenoids and sesquiterpenoids.

(LL 4258)

AUGUST 2017

Sub. Code: 4258

**B.PHARM. DEGREE EXAMINATION
SECOND YEAR**

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) Define and classify heterocyclic compounds.
b) Explain the structure, reactivity, preparation and chemical reactions of Pyridine.
2. a) Define and classify alkaloids with examples.
b) Elucidate the structure of Ephedrine.

II. Write notes on:

(8 x 5 = 40)

1. Explain in detail the criteria for a compound to be optically active.
2. Discuss about asymmetric synthesis with examples.
3. Explain the Cahn Ingold and Prelog system sequence rule for assigning configuration of optical isomers.
4. Explain inter relation between menthol, thymol and camphor.
5. Explain the mechanism and give the applications of Reduction using hydrazine.
6. Give the i) Fischer indole synthesis ii) Skraup synthesis of Quinoline.
7. Classify Flavanoids and discuss about the chemistry of Hesperidin.
8. Explain the Chemistry and pharmacological activity of digitalis glycosides.

III. Short answers on:

(10 x 2 = 20)

1. Write the structure and medicinal uses of Vitamin A and Vitamin D.
2. Define and give an application of Beckmann rearrangement.
3. Define Stereospecific and stereoselective synthesis.
4. Give any two applications of Lead tetra acetate.
5. What is Birch reduction? Give an example.
6. Write the formylation reactions of Pyrrole.
7. Write any two synthesis of Acridine.
8. List out the metal hydrides used in reduction reactions and give an application of Lithium aluminium hydride.
9. Write various conformations of cyclohexane.
10. Write the structure and medicinal uses of Citral and Caffeine.

THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

(LM 4258)

FEBRUARY 2018

Sub. Code: 4258

B.PHARM. DEGREE EXAMINATION

SECOND YEAR

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. Explain the following reactions as synthetic tools:
 - a) Oxidation with lead tetra acetate and periodic acid.
 - b) Beckmann rearrangement and Schmidt rearrangement.
2. a) Define and classify terpenoids with example.
b) Write the chemistry of alpha terpineol.

II. Write notes on:

(8 x 5 = 40)

1. Explain the various elements of symmetry with example.
2. Explain the chemistry of caffeine.
3. Discuss on tetrahedral carbon atom and stereochemistry of cyclic compounds.
4. Explain the sequence rules relating the R & S configuration.
5. Describe the common method of isolation of alkaloid.
6. Write the electrophili reaction of pyrrole.
7. Explain the chemistry of Vitamin A.
8. Discuss the pharmacological activity of atropine and related alkaloids.

III. Short answers on:

(10 x 2 = 20)

1. What is clemmenson's reduction?
2. Define Stereomutation.
3. Mention the medicinal uses of Vitamin B₆ and B₁₂.
4. Write the difference between enantiomer and diastereomer.
5. Give any two important reactions of isoxazole.
6. What is Walden inversion?
7. Define glycosides and name any two glycosides.
8. Write the structure and uses of papaverine.
9. Give the structure and uses of menthol and thymol.
10. Write the structural difference between theophyllin and theobromine.

THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

(LN 4258)

AUGUST 2018

Sub. Code: 4258

**B.PHARM. DEGREE EXAMINATION
SECOND YEAR
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY**

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on: **(2 x 20 = 40)**

1. a) Explain the various methods of determination of configuration of geometric isomers.
b) Give important methods of preparation and chemical reactions of indole.
2. a) Discuss the chemistry of camphor.
b) Explain the chemistry of Vitamin B₂ and Folic acid.

II. Write notes on: **(8 x 5 = 40)**

1. Explain the method of resolution by formation of diastereomers with examples.
2. Explain optical isomerism in Meso Tartaric acid.
3. Discuss the chemistry of pyrimidine.
4. Write the important reactions of quinoline.
5. Write the chemistry and uses of alpha-pinene.
6. Explain the basic ring system and nomenclature of steroid nucleus.
7. Discuss the inter-relationship between caffeine, theophylline and theobromine.
8. Discuss the chemistry of Digoxin.

III. Short answers on: **(10 x 2 = 20)**

1. What is conformational analysis?
2. Define relative configuration and absolute configuration.
3. Give two uses of mercuric acetate.
4. What is Darzen's reaction?
5. Give two methods of synthesis of thiazole.
6. Write any two reactions of Pyrazole.
7. Classify terpenoids.
8. What are sennosides?
9. Write the tautomers of uric acid.
10. Write the structure of Vitamin B₆ and its uses.

THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

(LO 4258)

FEBRUARY 2019

Sub. Code: 4258

**B.PHARM. DEGREE EXAMINATION
SECOND YEAR**

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on: (2 x 20 = 40)

1. Define and classify alkaloids with examples. Discuss the chemistry and pharmacological activity of Atropine.
2. a) Write in detail about asymmetric synthesis.
b) What is Walden Inversion explain in detail the mechanism of reaction and factors affecting Walden inversion?

II. Write notes on: (8 x 5 = 40)

1. Explain conformational analysis of ethane.
2. Explain the Schmidt rearrangement with mechanism.
3. Describe the chemistry of digitoxin.
4. Describe the chemistry and uses of Vitamin-E.
5. Write a brief note on catalytic hydrogenation.
6. Synthesis of thiophene.
7. Compare and contrast Clemmenson's reduction and Wolf Kishner reduction.
8. Explain the chemistry of Lanatosides.

III. Short answers on: (10 x 2 = 20)

1. Define racemic modification.
2. Define diastereomers.
3. Write the definition and reaction for Birch reduction.
4. Give any two methods of preparations of Imidazole.
5. What is metal hydride reduction? Give an example.
6. What are Terpenoids? Give two examples.
7. Write any two uses of Selenium oxide.
8. Define glycosides, give examples.
9. Write the structure of i) Vitamin - D ii) Vitamin – B₁.
10. Give any two reactions of Furan.

THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

(LP 4258)

AUGUST 2019

Sub. Code: 4258

B.PHARM. DEGREE EXAMINATION

SECOND YEAR

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) What are Cardiac glycosides? Explain the stereo chemistry of Cardiac glycosides.
b) Elucidate the structure of Ephedrine.
2. a) Explain the nomenclature of heterocyclic compounds.
b) Explain the structure, reactivity, preparation and chemical reaction of pyridine.

II. Write notes on:

(8 x 5 = 40)

1. Explain the electrophilic substitution reaction of pyrrole.
2. Explain the chemistry and use of Vitamin – K.
3. Give the reaction and mechanism of Meerwein-Ponndorf reduction.
4. Structural elucidation of uric acid.
5. Discuss about the chemistry of hesperidin.
6. Explain the chemistry and constitution of Papaverine.
7. Chemistry of Vitamin - A.
8. Explain the chemistry and constitution of citral.

III. Short answers on:

(10 x 2 = 20)

1. Synthesis of Acridine.
2. Two applications of Lead Tetra acetate.
3. Define Enantiomers.
4. Two important reactions of Oxazole.
5. Cis – trans Isomerism.
6. Define plane of symmetry.
7. Structure and Numbering of Phenothiazine.
8. What is 1, 3 diaxial interaction?
9. Define stereo-selective synthesis.
10. Define Isomers and Isomerism.

THE TAMIL NADU Dr.M.G.R. MEDICAL UNIVERSITY

(LQ 4258)

FEBRUARY 2020

Sub. Code: 4258

B.PHARM. DEGREE EXAMINATION

SECOND YEAR

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) Explain stereochemistry of cyclic compounds.
b) Explain various methods of determination of configurations of geometric isomers.
2. a) Discuss the chemistry and uses of ephedrine.
b) Explain the chemistry and medicinal uses of Vitamin B₁ and Vitamin E.

II. Write notes on:

(8 x 5 = 40)

1. Define and explain the pharmaceutical importance of chirality.
2. Explain conformational analysis of 1, 2 dichloroethane.
3. Explain various methods of resolution of racemic modification.
4. Discuss the chemistry of imidazole.
5. Explain the chemistry and uses of thymol.
6. Discuss the chemistry of lanatoside.
7. Explain the chemistry of uric acid.
8. Explain metal hydride reduction as synthetic tools.

III. Short answers on:

(10 x 2 = 20)

1. What is Fischer projection?
2. Define asymmetric synthesis.
3. What is E-Z system of nomenclature?
4. Write the structure and medicinal uses of camphor.
5. What are sennosides?
6. Write the structure and uses of folic acid.
7. Write the structure and medicinal uses of caffeine.
8. Give the medicinal uses of flavonoids.
9. What is periodic acid oxidation?
10. Give the synthetic uses of hydrazine and its derivatives.

THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

[LR 1220]

DECEMBER 2020

Sub. Code: 4258

(AUGUST 2020 EXAM SESSION)

B. PHARMACY DEGREE EXAMINATION

SECOND YEAR – (Regulation from 2009-2010)

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. What are Conformational Isomers? Explain the conformations of monosubstituted and disubstituted cyclohexanes.
2. a) Classify heterocyclic compounds and explain the nomenclature.
b) Explain the methods of preparation and chemical properties of Quinoline.

II. Write notes on:

(8 x 5 = 40)

1. Explain Asymmetric synthesis with examples.
2. Write the electrophilic substitution reactions of Thiophene.
3. Describe the chemistry and uses of Vitamin B6.
4. Classify Flavonoids. Explain the chemistry of Hesperidin.
5. What happens when:
 - a) Furan is treated with Maleic anhydride.
 - b) Thiophene is reacted with Acetyl chloride and stannous chloride.
 - c) Pyridine is nitrated with nitrating mixture.
 - d) Quinoline reacts with Methyl iodide.
6. Explain the chemistry of Sennosides.
7. Give the reaction and mechanism of Schmidt rearrangement.
8. Explain the modern theory of double bonds.

III. Short answers on:

(10 x 2 = 20)

1. Write any two methods of preparation of Pyrimidine.
2. Explain the reduction reaction with Hydrazines.
3. Mention any two uses of Sodium borohydride.
4. Give an identification test for Atropine.
5. Write any one method of preparation of Phenothiazine.
6. Write the structure of Vitamin D and Vitamin K and mention its uses.
7. What is Stereomutation?
8. Write any two reactions of Pyrazole.
9. Give any two oxidative uses of periodic acid.
10. Write the structure of α -pinene and its uses.

[BPHARM 0921]

SEPTEMBER 2021
(FEBRUARY 2021 EXAM SESSION)

Sub. Code: 4258

B. PHARMACY DEGREE EXAMINATION
SECOND YEAR – (NON-SEMESTER)
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY
Q.P. Code: 564258

Time: Three hours

Answer ALL Questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. What are Purines? Elucidate the structure of Uric acid. Discuss the interrelation between caffeine, theophylline and theobromine.
2. Explain the mechanism and applications of
 - a) Birch reduction
 - b) Meerwein Ponndorf reduction.

II. Write notes on:

(8 x 5 = 40)

1. Explain the chemistry of Vitamin E.
2. Give reasons for the following:
 - a) Explain why the electrophilic substitution reactions of Pyrrole predominantly occur at 2nd position than 3rd position.
 - b) Explain why Pyrrole and Furan is more reactive than Thiophene.
3. What is resolution of Racemic modification? Explain the method of resolution by biochemical separation.
4. Define optical activity. What are the various optical isomers possible in compounds containing two different chiral carbons?
5. Explain Fischer Indole synthesis.
6. Explain Cahn Ingold and Prelog system for assigning configuration based on sequence rules.
7. Explain the chemistry and pharmacological actions of Digitoxin.
8. Explain the chemistry of Camphor.

III. Short answers on:

(10 x 2 = 20)

1. What is Darzen's reaction?
2. Distinguish between Enantiomers and Diastereomers.
3. Why nitrobenzene is mostly used as oxidizing agent in Skraup synthesis.
4. Mention two uses of Lithium aluminum hydride.
5. Give the structure and uses of Folic acid.
6. Give two methods for the synthesis of Imidazole.
7. Write any two reactions of Furan.
8. What is Fischer projection?
9. Explain briefly D and L system of nomenclature.
10. Write the structure and use of Atropine.

**B.PHARMACY DEGREE COURSE (NON-SEMESTER EXAMINATIONS)
SECOND YEAR (Regulation 2009-2010)**

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) What is Asymmetric synthesis? Explain the types of asymmetric synthesis.
b) Explain Optical activity in Compounds containing two different chiral carbon atoms.
2. Define and classify heterocyclic compounds with examples. Explain the nomenclature of heterocyclic compounds. Write the methods of preparation and reactions of Quinoline.

II. Write notes on:

(8 x 5 = 40)

1. Describe the stereochemistry of cyclic compounds.
2. What is conformational analysis? Explain the conformational analysis of cyclohexane.
3. Explain lead tetra acetate oxidation as a synthetic tool.
4. Discuss the chemistry and uses of Menthol.
5. Explain the chemistry of Atropine.
6. Describe the chemistry of Sennosides.
7. Discuss the chemistry of Uric acid.
8. Explain the modern theory of double bonds in geometrical isomerism.

III. Short answers on:

(10 x 2 = 20)

1. What is Clemmenson's reduction?
2. Write any two reactions of Indole.
3. Giving an example and explain plane of symmetry.
4. Write any two reactions of Imidazole.
5. What is Stereoselective synthesis?
6. Write the structure of Papaverine and Ephedrine.
7. Give the structure and numbering of Digoxin.
8. Define Racemic modification.
9. Sketch the structures of Caffeine and Theobromine.
10. Can concentrated acids be used for electrophilic substitution reactions of Pyrrole? Justify your answer.

B. PHARMACY DEGREE EXAMINATION
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Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) What is Racemic modification? Write any two methods of preparation of Racemic modification.
b) Explain the different methods that are utilized for resolution of Racemic modification.
2. Describe the methods of preparation and chemical reactions of
a) Indole b) Quinoline.

II. Write notes on:

(8 x 5 = 40)

1. Describe the stereochemistry of cyclic compounds.
2. What is conformational analysis? Explain the conformational analysis of cyclohexane.
3. Explain the reaction and mechanism of Beckman rearrangement.
4. Discuss the chemistry and uses of Menthol.
5. Explain the chemistry of Citral.
6. Describe the chemistry of Folic acid.
7. Discuss the chemistry of Uric acid.
8. Explain the modern theory of double bonds in geometrical isomerism.

III. Short answers on:

(10 x 2 = 20)

1. What is Stereomutation?
2. Write any two reactions of Indole.
3. Give the synthetic applications of Mercuric acetate.
4. Write any two reactions of acridine.
5. What is Stereoselective synthesis?
6. Write the structure of Papaverine and Ephedrine.
7. Give the structure and uses of Vitamin B₁.
8. Define alkaloids and give examples.
9. Sketch the structures of Caffeine and Theobromine.
10. What is catalytic hydrogenation?

**B.PHARMACY DEGREE COURSE (NON-SEMESTER EXAMINATIONS)
SECOND YEAR (Regulation 2009-2010)**

PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY

Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. (a) Explain the structural elucidation of Atropine and its uses.
(b) Write the chemistry of Camphor.
2. Explain the mechanism and application of
(a) Meerwein-Ponndorf reduction.
(b) Beckmann Rearrangement.

II. Write notes on:

(8 x 5 = 40)

1. Explain the chemistry of Vitamin D.
2. Explain the chemistry of camphor.
3. Explain the interrelationship between Theophylline, Theobromine and caffeine.
4. Write the structural elucidation of uric acid.
5. Explain sequence rule.
6. Explain the various elements of symmetry with example.
7. Define flavonoids and write the chemistry of Hesperidin.
8. Write a note on Walden Inversion.

III. Short answers on:

(10 x 2 = 20)

1. Define Cis-Trans isomerism with example.
2. Write the structure and uses of menthol and thymol.
3. Write the structure and uses of papaverine.
4. Define alkaloids with example.
5. Define Terpenoids with example.
6. Write the synthesis of Imidazole.
7. Give any two reactions of Furan.
8. Write the structure and use of Vitamin A.
9. What is Darzens' Reaction?
10. What is metal hydride reduction reaction?

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SECOND YEAR (Regulation 2009-2010)
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Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. (a) What are vitamins and classify them with example?
(b) Explain the structural elucidation of Vitamin A.
2. (a) What are alkaloids? Classify them with example.
(b) Write a note on Resolution of Racemic modification.

II. Write notes on:

(8 x 5 = 40)

1. Explain the chemistry of uric acid.
2. Explain the sequence rule.
3. Explain the various elements of symmetry with examples.
4. Explain the mechanism and application of clemmensen reduction.
5. Write the synthesis and properties of pyrrole.
6. Write the Interrelationship between caffeine, Theophylline and Theobromine.
7. Write a note on Walden Inversion.
8. Explain the chemistry of Hesperidin.

III. Short answers on:

(10 x 2 = 20)

1. Define glycosides with example.
2. Define Terpenoids with example.
3. Define stereo mutation.
4. What is Birch Reduction?
5. Write any two reactions of furan.
6. Write the structure and uses of digitoxin.
7. Write the synthesis of pyridine.
8. Define Enantiomers.
9. What is Cis-Trans isomerism?
10. Write the structure and use of Vitamin B₁.

**B.PHARMACY DEGREE COURSE (NON-SEMESTER EXAMINATIONS)
SECOND YEAR (Regulation 2009-2010)
PAPER III – ADVANCED PHARMACEUTICAL ORGANIC CHEMISTRY**

Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) Explain the E, Z nomenclature of geometrical isomerism.
b) Explain the methods for the determination of configuration of geometrical isomerism.
2. Write the methods of preparation and electrophilic substitution reactions of indole.

II. Write notes on:

(8 x 5 = 40)

1. Explain the optical isomerism in meso tartaric acid.
2. Give reasons for the following:
 - a) Explain why pyridine is much lesser reactive than benzene towards electrophilic reagents.
 - b) Explain why pyridine undergoes electrophilic substitution at 3 position and nucleophilic substitution at 2 –and 4- position.
3. Explain Beckmann rearrangement with mechanism.
4. Explain the elements of symmetry.
5. Discuss the chemistry of Vitamin B 2.
6. Explain the chemistry of ephedrine.
7. Give the preparation and reactions of isoquinoline.
8. Explain the chemistry of citral.

III. Short answers on:

(10 x 2 = 20)

1. What is Schmidt rearrangement?
2. What is Clemmenson reduction?
3. Write two reactions of isoxazole.
4. What is asymmetric synthesis?
5. Write any two uses of lead tetra acetate.
6. What is stereospecific synthesis?
7. What are conformational isomers?
8. Write the structure of Vitamin B 1 and mention its uses.
9. Give the structure and uses of menthol and thymol.
10. What is catalytic hydrogenation?

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Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. a) What is cis-trans isomerism? Explain the E-Z system of nomenclature.
b) Discuss the methods for the determination of configuration of geometrical isomers.
2. a) Define and classify alkaloids with examples. Discuss the chemistry of ephedrine.
b) Describe the chemistry and uses of vitamin D and vitamin E.

II. Write notes on:

(8 x 5 = 40)

1. Write a note on Walden inversion.
2. Explain the elements of symmetry with suitable examples.
3. Write the reaction, mechanism and applications of Beckmann rearrangement.
4. Explain the electrophilic substitution reactions.
5. Outline the preparation and the important chemical reactions of phenothiazine.
6. Discuss the relationship between menthol and thymol.
7. Describe the chemistry of digitoxin.
8. Explain the mechanism and applications of Birch reduction.

IV. Short answers on:

(10 x 2 = 20)

1. Write any two uses of selenium oxide.
2. Give any two methods of preparation of pyrimidine.
3. Pyridine is more basic than pymole. Give reason.
4. Classify terpenoids.
5. Write the structure and uses of i) Atropine and ii) Papaverine.
6. Define and classify glycosides.
7. What is Darzen's reaction?
8. Write the structures of i) uric acid and ii) theophylline.
9. Name any one method used for the reduction of carbonyl compounds to hydrocarbons and give the reaction for the same.
10. What are flavonoids?

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Q.P. Code: 564258

Time: Three hours

Answer ALL questions

Maximum: 100 Marks

I. Elaborate on:

(2 x 20 = 40)

1. Define and classify Hetero cyclic compounds with examples. Explain the nomenclature of Hetero cyclic compounds. Write the methods of preparation and reaction of Quinoline.
2. (a) Write in detail about determination of configuration of Geometrical Isomerism.
(b) Write notes on Beckmann rearrangement.

II. Write notes on:

(8 x 5 = 40)

1. Chemical reaction of Indole.
2. Sequence rule for assigning configuration of optical isomers.
3. Write the application of,
 - a) Lithium Aluminium Hydride.
 - b) Selenium Oxide.
4. Explain chemistry of Vitamin E.
5. Explain the chemistry of Uric acid.
6. Explain about Schmidt rearrangement.
7. Discuss the chemistry of Ephedrine.
8. Describe the common method of isolation of Alkaloids.

III. Short answers on:

(10 x 2 = 20)

1. What is Walden inversion?
2. Medicinal uses of vitamin B6 and B12.
3. Write the difference between Theophylline and Theobromine.
4. What are Sennosides?
5. What is Fischer Projection?
6. What is Clemmensen's reduction?
7. What do you mean by Centre of symmetry?
8. Define and classify Terpenoids.
9. What is Chirality?
10. What is stereo selective synthesis?
